

Determination of the High-temperature Structures of Tugtupite using Synchrotron Radiation and Rietveld Refinements

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Beamline(s): X7B

The structural behaviour of tugtupite, (ideally $\text{Na}_8[\text{Al}_6\text{Si}_6\text{O}_{24}]\text{Cl}_2$), a member of the sodalite-group of minerals, was determined by using *in-situ* synchrotron radiation X-ray powder diffraction data ($\lambda = 0.91997 \text{ \AA}$) at room pressure and temperatures ranging from 38 to 985°C on heating. The sample was heated at a rate of about 9.5°C/min. and X-ray traces were collected at intervals of about 16°C. Eighteen structures in the temperature range were refined using the GSAS Rietveld refinement program. The variation of the volume with temperature is shown (Figure). Simultaneous differential scanning calorimetry (DSC) and thermogravimetry (TG) data were also obtained on tugtupite using a Netzsch STA 449C simultaneous TG-DSC instrument. Data were collected at a heating rate of 10°C/min. from 25 to 950°C. The peaks in the thermal curves are correlated with changes in the structural parameters. The unit-cell parameters for tugtupite increase quite smoothly and non-linearly to 985°C. In tugtupite, large displacement parameters occur for the Na and Cl atoms, and the Na - Cl bond expands with temperature. This forces the Na atoms toward the plane of the framework six-membered rings, which cause the framework tetrahedra to rotate. The framework TO_4 tetrahedra distort with temperature, with the distortions following the sequence $\text{Si} < \text{Al} < \text{Be}$, which parallels the T - O bond strengths. This mechanism initially causes a high rate of expansion in tugtupite. If the Na atom reaches the plane of the six-membered ring, the expansion will be retarded, but tugtupite melts before this occurs. Tugtupite begins to lose NaCl at about 844°C (the total mass change to 950°C is -1.6%), and melts at about 1029°C.

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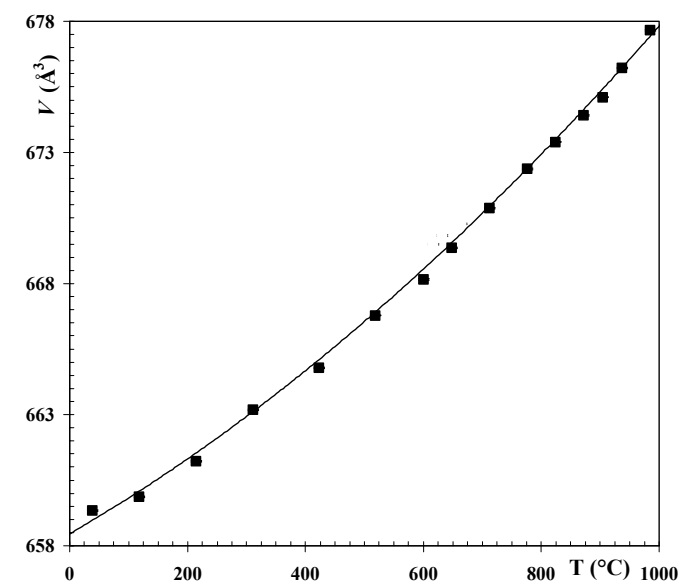


Figure: Variation of volume with temperature for tugtupite.